

# Effect of light on quality of milk

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Received: 01.05.2017; Revised: 04.10.2017; Accepted: 18.10.2017

■ **ABSTRACT :** Light-exposure is a very much chronicled issue in the investigation of the nature of milk. Milk subjected to light amid transport, handling, and storage can create undesirable off-flavors and fragrances, because of riboflavin prompted breakdown of proteins and lipids. In spite of much review into these marvels, most work to date has been completed with incandescent or fluorescent light sources. The present day production network and retail condition is advancing, and right now the dairy business is seeing numerous providers embracing LED lighting in show cases, basically as a cost sparing and maintainability measure, because of lower running expenses. LED lighting produces a highly modified range of light than the fluorescent frameworks at present supported. The nearness of milk fat seems to ensure against vitamin A debasement in liquid items, however unfavorably influences the flavor nature of milk after presentation to light.

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■ **KEY WORDS:** Vitamin A, Light oxidation, Fortified milk, LED lights, Light-exposed flavor

■ **HOW TO CITE THIS PAPER :** Khan, Niha and Saxena, Karunesh 2017). Effect of light on quality of milk. *Asian J. Home Sci.*, 12 (2) : 427-431, DOI: 10.15740/HAS/AJHS/12.2/427-431.

Light exposure can adversely influence the wholesome esteem and flavor nature of liquid milk items. Both normal and artificial light, especially in wavelengths running from 420 to 520 nm, can actuate quality deformities (Bosset *et al.*, 1994). Previous reports have detailed light-induced chemical reactions that result in vitamin A degradation and light-oxidized flavor defects (Fanelli *et al.*, 1985; Fellman *et al.*, 1991; Cladman *et al.*, 1998) following 2 to 4 h of exposure in a lighting system that simulated commercial display cases. (Hansen *et al.*, 1975). Gaylord *et al.* (1986) showed a lessening in the vitamin A substance of entire milk, diminished fat, low fat, and nonfat milk presented to 2000 lx for 4 h. Also, packaging type and size impacts vitamin A losses and lessening of flavor quality. Of milk bundled in polyethylene terephthalate (PET), PET with an UV

blocker, low-thickness polyethylene (LDPE) or high-thickness polyethylene (HDPE), items in HDPE had the biggest vitamin A losses (Cladman *et al.*, 1998). The dominant part of liquid milk items as of now are packed in containers made of paperboard or HDPE plastic. Paperboard offers insurance from light presentation, yet HDPE compartments transmit light.

Light-exposed flavor is a long-studied phenomenon in the quality of milk (Brown, 1899). At the point when presented to daylight or bright light amid transportation and capacity (Jung *et al.*, 1988) the vitamins, proteins and lipids of milk experience various responses, some of which may prompt the advancement of undesirable off-flavors and fragrances, or the breakdown of supplements. Light exposed flavor is frequently labeled as "light-oxidized", not generally rectify in milk, because of light

additionally activating protein breakdown, however because of the inevitable light instigated oxidation of lipid parts, which normally happen after the debasement of such proteins, these terms are regularly connected synonymously.

Light introduction negatively affects the impression of milk. Most buyers lean toward the essence of non-light presented tests to light subjected examples. In a review by White and Bulthaus, 1982, buyers were made a request to assess milk presented to direct daylight for 20 minutes and milk with no light presentation. Of the 278 purchasers that distinguished the odd specimen, 63% of shoppers favored the non-light oxidized milk, with 27% picking the light oxidized example and 10% having no inclination. White and Bulthaus inferred that the normal purchaser may not grumble altogether about light oxidized flavor in milk however would convey what needs be by acquiring less milk. Thus, when utilizing a dairy item judging board, Barnard, 1973, detailed that 23.9% of milk tests were appraised reasonable for poor because of light-uncovered off-flavor and presumed this off-flavor adds to a decrease in milk utilization.

### **Light effect on Riboflavin, a naturally present vitamin in milk :**

While light-oxidation is a mind boggling process including various parts, it commonly starts with the photo excitation of riboflavin (Aurand *et al.*, 1977) a naturally present vitamin in milk, or other photosensitive segments contained inside (Wold *et al.*, 2005). Riboflavin has two features amid light-oxidation, it assimilates light; and goes about as a bearer of hydrogen and electrons for decrease of different substances. Riboflavin is a powerful light-safeguard and photo sensitizer because of its conjugated twofold bond structure. As light vitality is assimilated, electrons are supported to higher vitality levels, conveying the photo sensitizer to the singlet-energized state. This may happen along two diverse pathways, Type I or Type II. In Type I, the sensitizer cooperates straightforwardly with another atom, bringing about the creation of free radicals. In Type II, vitality is exchanged from the energized triplet sensitizer to oxygen, creating the energized singlet condition of oxygen (Bradley and Min, 1992). Riboflavin tends to support the Type I pathway despite the fact that it might experience the Type II response.

Contrasting reports exist on the excitation

wavelength with the most potential for harm. Webster *et al.* (2009) detailed that riboflavin has a tendency to wind up noticeably energized at wavelengths of 400, 446 and 570 nm. Wavelengths of 350 - 520 nm were harming to riboflavin, with wavelengths in the 415 – 455 nm area being the most dangerous. Allen and Park (1979) expressed that it is important to farthest point light wavelengths in the vicinity of 440 and 480 nm, and that the wavelength of 450 nm was the most dangerous to riboflavin.

At the point when riboflavin ends up noticeably energized it aids the breakdown of proteins and lipids, creating aggravates that produce the off-flavors unmistakable of light-introduction. There are two principle terms used to order off-flavors related with light-introduction of milk; activated flavor and oxidized flavor. Enacted flavor is made by the breakdown of serum proteins in the production of actuated flavor is the light-incited breakdown of the amino corrosive methionine, which just seems to happen within the sight of riboflavin (Dimick, 1982). Upon introduction to light methionine can debase into methional, which can then corrupt further into dimethyl disulfide, methyl mercaptan or other sulfur mixes (Finley and Shipe, 1969). These mixes are profoundly connected with the improvement of initiated off-flavor.

Perceptible light-uncovered off-flavors in milk may create after a short measure of presentation to light. In a review directed by Chapman *et al.* (2002), customers and prepared specialists were made a request to assess 2% milk. Untrained specialists could identify season deserts in as meager as 54 minutes to 2 hours of introduction, while the limit of the prepared specialists was as short as 15 to 30 minutes .

### **Relation between light, sensory changes, the nutritional quality and visual aspect of milk :**

Notwithstanding off-flavors, light-oxidation makes hindering wholesome misfortune from the pulverization of vitamins A, C, D and E (Kim and Mahony, 1998). Ascorbic corrosive (vitamin C) contends with fats over oxygen, making it quickly corrupt (Kon *et al.*, 1936). The rate of ascorbic corrosive disintegration develops as light force increments. Riboflavin likewise debases from similar wavelengths that energize it. Light presentation will likewise change the shading organization of milk. Milk presented to direct daylight can turn dark

colored because of the breakdown of milk proteins (Toba *et al.*, 1980). A recent report found that instrumental readings of greenness and yellowness diminished, moving towards red and blue individually, a comparable outcome to Mestdagh *et al.* (2005). It was in this manner estimated that the shading change was brought about by the debasement of riboflavin, which is yellow-green hued,  $\beta$ -carotene and vitamin A atoms.

The normal parts of milk can impact the advancement of light-uncovered flavor. A few vitamins go about as cancer prevention agents to keep the advancement of off-flavors or pulverization of supplements. Ascorbic corrosive can be a viable quencher for enacted oxygen species and can keep the debasement of vitamin A analogs (Karatapanis *et al.*, 2006) and vitamin E subsidiaries. At the point when consolidated with  $\alpha$ -tocopherol (Vitamin E), ascorbic corrosive can constrain the era of light oxidized flavor,  $\alpha$ -tocopherol can likewise forestall lipid oxidation to a degree when covered onto the bundling material of milk (Lee *et al.*, 2004). Fat substance likewise impacts the light-instigated oxidation prepare. Fat can at first shield milk from vitamin misfortune and flavor surrenders (Whited *et al.*, 2002) and the light-corruption rate of riboflavin is slower as the fat substance of the milk increments.

A few stages might be taken to keep the quality issues made by light introduction, with the most straightforward to keep light from illuminating the milk. Bundling material significantly affects the improvement of light-uncovered flavor. By and large, plastic compartments have expanded the occurrences of light-oxidized flavor in milk (Barnard, 1973). High-thickness polyethylene (HDPE), a typical material in milk bundling, is not as compelling in ensuring the milk as paperboard (Wold *et al.*, 2005) or polyethylene terephthalate (PETE) (Van Aardt *et al.*, 2001). This is likely because of the light penetrability of HDPE when contrasted with PETE or different materials (Cladman *et al.*, 1998). Tinted or pigmented material can demonstrate viable in keeping the advancement of off-flavor mixes (White and Bulthaus, 1982 and Herreid *et al.*, 1952). In a review directed by Van Aardt *et al.* (2001) golden shaded polyethylene terephthalate (PETE) was more successful at blocking wavelengths beneath 450 nm and had less oxidized flavor than clear PETE holders. In any case, not all bundling techniques offer to shoppers. Despite

the fact that buyers would buy shaded holders if the cost is comparable to translucent compartments, they commonly favor white or cream hued holders for milk (Allen and Parks, 1979). Ultimately, cancer prevention agents, for example, the previously mentioned vitamins or plant polyphenols might be utilized to keep the advancement of light-uncovered deformities (Van Adardt *et al.*, 2001; Becker *et al.*, 2005 and Lemieux and Simard, 1994).

### Conclusion :

The changing wavelength profiles made by the innate mechanics of each light source is likely the reason for any variety in tactile properties of milk. In further reviews, both human tactile and gas chromatography examination would be useful to research the connection between's the recognition or centralization of particular fragrance volatiles, and the technique for light introduction. There exists a judicious probability that LED lights have a contrasting effect on the tangible nature of milk when contrasted with introduction from fluorescent lighting. As the larger part of buyers favor the essence of light ensured milk to light-uncovered milk (Martin *et al.*, 2016), exchanging glaring lights with LEDs for vitality effectiveness may influence the milk's flavor, and further impact offers of dairy items. Lessening in light-uncovered off-flavor might be particularly essential in empowering kids and youthful grown-ups to drink milk, with the objective of eventually transforming them into future milk buyers, as youngsters have been appeared to hate the essence of light-uncovered milk. Despite the fact that LEDs are in effect energetically actualized as light sources, glaring lights may in any case be available in an extent of dairy cases.

Utilization of bundling materials that give optical lucidity through wavelength obstruction, as opposed to retention, as over-wraps can give another road to esteem included bundling. These bundling materials might be enhanced to square particular wavelengths that will secure milk enhance and take into account increased timeframe of realistic usability. In any case, it is as yet vague which wavelengths, and the level of transmission decrease of these wavelengths, are expected to completely ensure against light oxidation enhance generation. Bundling overwraps produced using luminous film diminished the sensory view of light-oxidized flavor in milk yet were not as proficient as the light-secured

treatment in lessening the generation of light oxidation season.

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